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A new and efficient conjugate addition of trimethyl and triethyl phosphites to 3 - ω -azidoacetylcoumarin (1) has been studied. The reaction proceeded smoothly at r.t. furnishing 1,2,3,4-triazaphosphole coumarin derivatives 4a,b in \sim 75% yields. Linear substituted triazoles 10a,b were also obtained from the reactions of 1 with α -keto ylides, acetyl- and benzoylmethylene triphenylphosphoranes. Contrary to these results, Wittig reaction was occurred when 1 was allowed to react with a-alkoxycarbonylmethylene- and cyanomethylenetriphenylphosphoranes 7c–e as well as with methylidene- and benzylidenetriphenylphosphoranes 8a,b resulting in the formation of the corresponding olefins either as an intermediate 14b or as final products 11a–c.

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INTRODUCTION

A considerable number of naturally occurring coumarins such as Muttalongin, Usthol, and 2,3-Auraptin were found to have strong antimicrobial and anticancer activities [1]. Also, coumarin dyes recently have found great interests because of their high potential in dye laser to achieve tunable blue-green light and are also used in enzyme determination; photobiological energy transfer processes fluorescent probe technique; and fluorescence whiteners in detergent products [2].

In continuation with our research program directed toward the utility of trivalent and pentavalent phosphorus reagents in synthesis of pharmaceuticals [3–12], we extend our precedent work on coumarins and relevant phosphor derivatives [13,14] to elaborate more novel coumarin derivatives of expecting pharmacological potenc. The methodology involved application of trialkyl phosphites and alkylidenephosphoranes to 3-azidoacetylcoumarin (1).

RESULTS AND DISCUSSION

The azide 1 reacted smoothly with trimethyl- 2a and triethyl phosphite 2b in absolute methylene chloride at room temperature and yielded, in each case, the tautomeric structure $4a,b = 4Aa,b$ in $\approx 75\%$ yield. Compound 4 can be viewed as derived from ring closure of the phosphazide 3, initially formed [15,16] with concomitant extrusion of the appropriate alcohol moiety. Further rearrangement through alkyl group shift led to the triazaphosphole products 4a,b (Scheme 1).

Compounds $4a,b$ showed $31P-NMR$ chemical shifts around δ 11.4 ppm confirming the presence of N-P-C linkage in a phosphole moiety and readily eliminate any formation of phosphates for which a signal at $\delta p = \pm$ 2–4 ppm would be expected. The IR spectrum of 4a revealed the presence of absorption bands at 1722 $(C=0, \text{ lactone}), 1633 (C=0, \text{acetyl}), 1265 (P=0, \text{$ free), and at 1055 $(P-O–CH₃)$ cm⁻¹. Moreover, the strong azide frequency present in the IR spectrum of 1 at 2093 cm^{-1} was absent in the IR spectra of 4. In the

¹H-NMR [17] spectrum of $4a$ (CDCl₃), the 4'-H of coumarin was observed as a singlet at δ 8.68 ppm. A doublet (3H, ${}^{3}J_{\text{P-H}}$ = 8.8 Hz) was observed at δ 3.66 due to OCH₃ protons attached to the phosphorus atom. A doublet (3H, ${}^{3}J_{\text{P-H}}$ = 8.3 Hz) was observed at δ 3.27 assigned to the $CH₃$ protons linked to the 3-N in the triazaphosphole ring. The singlet due to the methylene protons linked to the azide group present in the pmr spectrum of 1 at δ 4.73 ppm was absent in the spectrum of **4a**. Instead, a doublet $(1H, {}^{2}J_{P-H} = 23.4 \text{ Hz})$ was displayed at δ 5.16 due to 5-H (4a). Remarkably, the structure 4Aa (Scheme 1) cannot be overlooked since a weak broad signal was displayed at 3339 cm^{-1} in the IR spectrum assigned to an NH group; this signal appeared at δ 11.42 (br) in its pmr spectrum. It is noteworthy that the position of N-(H) hydrogen atom in triazoles such as structure 4 has been the subject of contradictory discussion $[18]$. Furthermore, the ³¹P-NMR chemical shifts; the presence of 5-H as a doublet of high coupling constant and 4'-H coumarin singlet, and the lack of a singlet due to the methylene group in the pmr spectrum, as well as the absence of the azido group in its ir spectrum, confirm the assigned structures $4 - 4A$ and rule out other alternative structures like 5 and 6.

Next, the study has been extended to investigate the interaction of the azide compound 1 with three types of stabilized Wittig reagents: $i - \alpha$ -ketomethylene-7a,b, ii- alkoxycarbonyl-methylene-7c,d, iii- cyanomethylenetriphenylphosphorane 7e; and two active ylides: i- methylidene- (8a) and benzylidenetriphenylphosphorane (8b).

When the azide 1 was treated with acetylmethylene-7a or benzoylmethylenetriphenyl-phosphorane (7b) in tetrahydrofuran (THF) at room temperature for 24 h, 1,2,3-triazoloactyl-coumarins 10a,b were formed in \sim 74% yield. Triphenylphosphine oxide (TPPO) was also isolated from the reaction medium. The triazole structure 10 was deduced from analytical and spectroscopic data. In the IR spectrum of 10a, the lactone carbonyl was observed at 1721 and that of the 3-acetyl carbonyl group observed at 1685 cm^{-1} . In its ¹H-NMR, the protons of methylene group were observed downfield as a singlet at δ 5.37 ppm.

As depicted in Scheme 2, a 1,3-dipolar addition of the azide 1 on the $P=C$ bond of the ylide 7a,b followed by elimination of TPPO from the cyclic intermediate 9 afforded N-1-substituted-5-methyl (or 5-phenyl)-1,2,3 triazoles 10a,b [19–23] . The formation of the triazoles from treating 1 with 7a,b adequately demonstrates the regiochemistry [23] of the reaction that leads to the exclusive formation of 1,5- instead of 1,4-disubstituted triazoles. Furthermore, the formation of 10 through either a two step addition, via the intermediate 9, or a

concerted cycloaddition, via the transition state 9A, is consistent with the observed result.

In the contrary to the behavior of trialkyl phosphites and α -keto ylides, the reactions of 1 with α -alkoxy carbonyl- and a-cyanomethylenetriphenylphosphoranes (7c–e), performed under similar conditions, followed the Wittig reaction, gave coumarin-azido-derivatives 11a–c in \approx 70% yields. Compounds 11a–c were obtained as Zisomers predominantly based of spectroscopic data and due to steric demand of the transition states to form the oxaphosphetane 9A (Scheme 3) [24–27] .

The spectroscopic data of compounds 11a-c are in agreement with the assigned structure. In the EI mass

spectra, they showed the molecular ion peaks $(M^+ \%)$ that displayed the expected fragmentations of the corresponding spectrum. The most characteristic signals in their ¹H-NMR spectra are those of the methylene $(CH₂N₃)$ and the methinic (=CHY) protons, which appeared as a doublet and a triplet each with $^{4}J_{\text{H-H}}$ = 1.6 Hz at δ 4.44–4.23 and 6.08–6.28, respectively. The presence of $(=CHY)$ and $(CH₂N₃)$ moieties were also attested by signals at δ c 112-98, and 55–52 ppm in ¹³C-NMR spectra. These recorded data for the methine $(=CH)$ and the methylene protons could readily eliminate structure like 12, which would predict a doublet at \sim 3.5 and a triplet at \sim 7.3 ppm.

Further extension of this study to other ylides, application of methylidene- (8a) and benzylidene-triphenylphosphorane (8b) to 1 was investigated. When dimethylformamide (DMF) solution of an equivalent amounts of 1 and 8a, prepared in situ from its bromide salt 13a, in the presence of aqueous LiOH was stirred at rt for 12 h, the reaction afforded the Wittig product 14a (80%), on the basis of comparable spectroscopic arguments (Scheme 4).

On the other hand, the reaction of 1 with slight excess (10%) of 8b (prepared in situ from its chloride salts, 13b) under the same conditions gave the benzylvinyl derivative 15 (84%), which is an isomer of the Wittig product 14b (Scheme 4).

In summary, the condensation of trialkyl phosphites and alkylidenephosphoranes (resonance-stabilized- and active ylides) with 3-azidoacetylcoumarin resulted in an interesting spectrum of the reactivity. Furtherward, the structural products $(4a,b, 10a,b, 11a-c, 14a, and 15)$ obtained from the four studied reactions, however, indicated two positions in 1 that are susceptible to nucleophilic attack by the phosphorus reagents. The first position relates to an attack at the azide group; this addition was observed by trialkyl phosphites and α -keto ylides, which afforded, via 1:3-dipolar intermediates, 1,2,3,4-triazaphosphole and triazolylacetyl-coumarin derivatives, respectively. The second site of attack is concerned with the attack of the ylides 7c–e and 8a,b at the acetyl carbonyl carbon group, which led to the respective olefins.

Finally, the synthesized heterocyclic phosphole system 4a,b that included a phosphorus linked to a nitrogen atom are of great interest because this system is common to a diverse array of important biological molecules.

EXPERIMENTAL

General: Melting points were determined with open capillary tube on an Electrothermal (variable heater) melting point apparatus and were uncorrected. IR spectra were recorded on a Perkin–Elmer spectrophotometer model 297 using KBr disc. NMR spectra were measured with a JEOL E.C.A-500 MHz $(^{13}C: 125.7$ MHz, $^{1}H: 500$ MHz, $^{31}P: 202.4$ MHz) spectrometer. $31P-NMR$ spectra were recorded with H_3PO_4 (85%) as external reference. ¹H- and ¹³C-NMR spectra were recorded with trimethylsilane as internal standard in $CDCl₃$ or $DMSO$ d^6 . Chemical shifts (δ) are given in ppm. The mass spectra were performed at 70 eV on an MS-50 Kratos (A.E.I.) spectrometer provided with a data system. The appropriate precautions in handling moisture-sensitive compounds were observed. Materials and reagents were purchased from Aldrich Company. The substrate, 3-azidoacetycoumarin 1 was prepared according to the reported method [15] .

Reaction of 3-azidoacetylcoumarin (1) withtrialkylphosphites 2a,b. Preparation of compounds 4a and 4b. To a solution of $0.8 \text{ g } (3.5 \text{ mmol})$ of the azide 1 in $25 \text{ mL of absolute}$ CH_2Cl_2 at $0^{\circ}C$, 3.5 mmol of freshly distilled trimethyl- 2a or triethyl phosphite 2b were added dropwise with stirring. After the addition was complete, the reaction mixture was stirred at room temperature for 6 h, and the solvent was evaporated to dryness. The residue was washed several times with light petroleum $(40-60^{\circ}C)$ and crystallized from the proper solvent to give 4a or 4b, respectively.

3-[(4-Methoxy-3-methyl-4-oxido-4,5-dihydro-3>H-1,2,3,4 triazaphosphol-5-yl)-carbonyl]-2H-chomen-2-one (4a). This was obtained as straw yellow needles, 863 mg (77%); mp 170–172°C (cyclohexane); IR: v_{max} NH 3339 (4Aa), 2C=O 1722, 1633, P=O, free 1265, 1055 (P-O-C) cm⁻¹; ¹H-NMR (DMSO⁶): δ 3.27 (d, ³J_{P-H} = 8.3 Hz, 3H, NCH₃), 3.66 (d, ³J_{P-H} $= 8.8$ Hz, 3H, POCH₃), 5.16 (d, ²J_{P-H} = 23.4 Hz, 1H, 5-H, 4a), 7.32 (t, $J_{\text{H-H}} = 7.4$ Hz, 2H, 6'-H and 7'-H), 7.52 (d, $J_{\text{H-H}} = 7.8$ Hz, 2H, 5'-H and 8'-H), 8.68 (s, 1H, 4'-H), 11.42 ppm (br, 1H, NH, 4Aa); ¹³C-NMR (DMSO⁶): δ 177.4 (d, ²J_{P-C} = 5.8 Hz, $C=0$), 158.8 ($C=0$, lactone), 153.3 (9'-C), 147.7 (4'C), 134.3, 131.5, 126.4, 118.7, 116.2 (7'-C, 8'C, 5'-C, 6'-C, 10'-C), 133 (d, ¹J_{P-C} = 139 Hz, 5-C-P, **4Aa**), 120.2 (d, ³J_{P-C} = 4.4 Hz, 3²-C), 65.3 (d, ¹J_{P-C} = 141 Hz, 5-C-P), 55.1 (d, ²J_{P-C} = 7.5 Hz, POCH₃), 28.8 ppm $(d_1^2 J_{P-C} = 7$ Hz, N-CH₃); ³¹P-NMR (DMSO⁶): δ 11.4 ppm; ms: m/z (EI): 323 [M⁺ + 2, 80 %], 321 $[M⁺, 100\%]$, 306 (16), 291 (31), 252 (83), 228 (33), 227 (24), 213 (19), 188 (18), 173(43), 170 (10), 77 (18). Anal. Calcd. for $C_{13}H_{12}N_3O_5P$ (321.2): C, 48.61; H, 3.77; N, 13.08; P, 9.64. Found: C, 48.68; H, 3.82; N, 13.15; P, 9.59.

3-[(4-Ethoxy-3-ethyl-4-oxido-4,5-dihydro-3H-1,2,3,4-triazaphosphol-5-yl)-carbonyl]-2H-chromen-2-one (4b) This was obtained as yellow needles, 0.9 g (74%); mp 132-134°C (cyclohexane); IR: v_{max} NH 3384 (4Ab), 2C=O 1722, 1638, P=O, free 1258, 1088 (P-O-C) cm⁻¹; ¹H-NMR (DMSO⁶): δ 1.03–1.31 (2dt (m), 6H, NC.CH₃ and POC.CH₃), 3.99 (dq, $J_{\text{H-H}} = 6.4, {}^{3}J_{\text{P-H}} = 4.8 \text{ Hz}, 2\text{H}, \text{NCH}_2$, 4.09 (dq, ${}^{3}J_{\text{H-H}} =$ 6.4, ${}^{3}J_{\text{P-H}} = 4.8$ Hz, 2H, POCH₂), 5.09 (d, ${}^{2}J_{\text{P-H}} = 23.4$ Hz, 1H, 5-H, 4b), 7.35 (t, $J_{\text{H-H}} = 7.4$ Hz, 2H, 6'H and 7'H), 7.57 $(d, J_{H-H} = 7.8 Hz, 2H, 5'H and 8'H), 8.64 (s, 1H, 4'H), 11.14$ ppm (br, 1H, NH, 4Ab); ¹³C-NMR (DMSO⁶): δ 177.6 (d, ²J_P. $C = 5.8$ Hz, $C=O$), 158.8 (C=O, lactone), 154.7 (9'-C), 147.8 (4'C), 134.7, 130.6, 126.4, 118.3, 117.6 (7'-C, 8'-C, 5'-C, 6'- C, 10'-C), 136 (d, $^{1}J_{\text{P-C}} = 139$ Hz, 5-C-P, 4Ab), 120.4 (d, $^{3}J_{\text{P}}$ C_{C} = 4.4 Hz, 3'-C), 67.4 (d, ¹J_{P-C} = 144 Hz, 5-C-P), 62.5 (d, ²J_{P-C} = 7.5 Hz, POCH₂), 44.1(d, ²J_{P-C} = 7 Hz, N-CH₂), 19.6 (d, ${}^{3}J_{\text{P-C}} = 5.8$ Hz, POC.CH₃), 14.4 ppm (d, ${}^{3}J_{\text{P-C}} = 5.8$ Hz, N-C.CH₃); ³¹P-NMR (DMSO⁶): δ 10.8 ppm; ms: m/z (ei): $351 \text{ [m+ + 2, 26\%]}$, 349 (74) [m+] , 320 (17), 291 (37), 263(100), 252 (71), 227 (83), 213 (20), 188 (13), 173(53), 170 (15), 77 (18); Anal. Calcd. for $C_{15}H_{16}N_3O_5P$ (349.3): C, 51.58; H, 4.62; N, 12.03; P, 8.87. Found: C, 51.65; H, 4.57; N, 12.11; P, 8.91.

Reaction of 1 withketomethylenetriphenyphosphoranes7a,b. Preparation of compounds 10a and 10b. A mixture of azide 1 $(0.8 \text{ g}, 3.5 \text{ mmol})$ and acetyl- 7a $(1.3 \text{ g}, 3.5 \text{ mmol})$ or benzoylmethylenetriphenylphosphorane (7b) (1.1 g, 3.5 mmol) in 30 mL of dry tetrahydrofuran (THF) was stirred at rt for 24 h. After removing the solvent, the residue was chromatographed on silica gel to afford the triazoles 10a or 10b. Triphenylphosphine oxide (Ph₃PO) was also isolated.

3-[(5-Methyl-1H-1,2,3-triazol-1-yl)acetyl]-2H-chromen-2 one (10a). This was obtained as yellow crystals, 705 mg (75%); mp 158-160°C (MeCN); IR: v_{max} 2C=O 1721, 1685 cm⁻¹; ¹H-NMR (DMSO⁶): δ 2.38 (d, $J_{\text{H-H}} = 2.3$ Hz, 3H, 5-C.CH₃), 5.37 (s, 2H, C(O)CH₂), 7.34 (d, $J_{\text{H-H}} = 7.4$ Hz, 2H, 6'-H and 7'-H), 7.55 (d, $J_{\text{H-H}} = 7.8$ Hz, 2H, 5'-H and 8'-H), 7.58 (q, $J_{\text{H-H}} = 2.3$ Hz, 1H, 4-H), 8.65 ppm (s, 1H, 4'-H); ¹³C-NMR (DMSO⁶): δ 187.4 (C=O, acetyl), 161.2 (2'-C=O), 154.7 (9'-C), 148.7 (4'-C), 142.8 (5-C), 138.3 (4-C), 134.3, 131.5, 126.4, 118.7, 114.2 (7'-C, 8'-C, 5'-C, 6'-C, 10'-C), 121.6 (3'-C), 53.8 (CH₂, acetyl), 13.3 ppm (5-C-CH₃); ms: m/z (EI): 270 (27) $[M^+ + 1]$, 269 (22) $[M^+]$, 254 (26), 237 (100), 188 (63), 172 (53), 170 (15), 145 (18), 77 (16). Anal. Calcd. for $C_{14}H_{11}N_3O_3$ (269.3): C, 62.45; H, 4.12; N, 15.61. Found: C, 62.52; H, 4.06; N, 15.64.

3-[(5-Phenyl-1H-1,2,3-triazol-1-yl)acetyl]-2H-chromen-2 one (10b). This was obtained as yellow crystals, 855 mg (74%); mp 186-188°C (MeCN); IR: v_{max} 2C=O 1720, 1680 m^{-1} ; ¹H-NMR (DMSO⁶): δ 5.27 (s, 2H, C(O)CH₂), 7.24–7.37 (m, 3H, H-Ph and H-Ar), 7.46–7.66 (m, 6H, H-Ph and H-Ar), 8.02 (s, 1H, 4-H), 8.53 ppm (s, 1H, 4'-H); ¹³C-NMR (DMSO⁶): δ 186.7, 160.4 (2C=O), 154.7 (9'-C), 148.7 (4'-C), 142.8 (5-C),)134.3 (4-C), 131.5, 130.5, 129.2, 126.4, 125.7, 123.6, 123.1, 122.6, 118.7, 114.2 (C-Ph and C-Ar), 121.6 (3'-C), 53.8 (CH₂, acetyl); ms: m/z (EI): 333 (36) [M⁺ +2], 331(100) [M⁺], 303 (92), 289 (78), 288 (51), (26), 237 (100), 188 (63), 172 (53), 170 (15), 145 (18), 77 (14). Anal. Calcd. for $C_{19}H_{13}N_3O_3$ (331.3): C, 68.88; H, 3.95; N, 12.68. Found: C, 68.85; H, 4.01; N, 12.62.

Reaction of 1 with alkylidenetriphenyphosphoranes 7c– e.Preparation of compounds 11a, 11b, and 11c. A mixture of the azide 1 (0.8 g, 3.5 mmol) and 3.5 mmol of methoxycarbonyl- (7c), ethoxycarbonyl- (7d), or cyanomethylenetriphenylphosphorane (7e) in 30 mL THF was stirred at r.t. for 24 h. After removing the solvent, the residue was chromatographed on silica gel to afford $11a$, $11b$, or $11c$, respectively. Ph_3PO was also isolated.

Methyl (2Z)-4-azido-3-(2-oxo-2H-chromen-3-yl)but-2-enoate (11a). This was obtained as yellow crystals, $657 \text{ mg } (66\%)$; mp 165–167°C (CHCl₃); IR: v_{max} C-N₃ 2112, C=O (lactone) 1721, C=O (ester)1713, C=C (exocyclic) 1611 cm^{-1} ; ¹H-NMR (DMSO⁶): δ 3.73 (s, 3H, OCH₃), 4.23 (d, ⁴J_{H-H} = 1.6 Hz, 2H, H_2C-N_3), 6.28 [t, ${}^4J_{\text{H-H}} = 1.6$ Hz, 1H, $=CHC(0)$], 7.34 (t, $J_{\rm H-H} = 7.4$ Hz, 2H, 6'-H and 7'-H), 7.56 (d, $J_{\rm H-H} =$
7.34 (t, $J_{\rm H-H} = 7.4$ Hz, 2H, 6'-H and 7'-H), 7.56 (d, $J_{\rm H-H} =$ 7.8 Hz, $2H$, $5'$ -H and $8'$ -H), 8.45 ppm (s, 1H, 4'-H); NMR (DMSO⁶): δ 162.5, 161.8 (2C=O), 153.7 (9'-C), 146.7 $(3'-C—C=), 141.7 (4'-C), 133.3, 126.4, 125.6, 118.7, 114.2)$ $(7'-C, 8'-C, 5'-C, 6'-C, 10'-C), 113.2 (3'-C), 112.8$ [=CHC(O)], 56.5 (H₂C-N₃), 52.6 ppm (OCH₃, ester); ms: m/z (EI): 286 (9) $[M^+ + 1]$, 285 (17) $[M^+]$, 254 (11), 243 (100), 225 (32), 198 (28), 171 (43), 170 (19), 142 (10), 114 (8). Anal. Calcd for C₁₄H₁₁N₃O₄ (285.3): C, 58.95; H, 3.89; N, 14.73. Found: C, 59.01; H, 3.84; N, 14.67.

Ethyl (2Z)-4-azido-3-(2-oxo-2H-chromen-3-yl)but-2-enoate (11b). This was obtained as yellow leaflets, 710 mg (68%) ; mp 128–130°C (cyclohexane); IR: v_{max} N₃ 2102, C=O (lactone) 1720, C=O (ester) 1714, C=C (exocyclic) 1608 cm⁻¹;
¹H-NMP (DMSO⁶): δ 1.23 (f) 3H, L₂₂ = 6.8 Hz, OC CH₂) H-NMR (DMSO⁶): δ 1.23 (t, 3H, $J_{\text{H-H}} = 6.8$ Hz, OC.CH₃), 4.08 (q, $J_{\text{H-H}} = 6.8$ Hz, 2H, OCH₂, ester), 4.39 [d, 2H, ⁴ $J_{\text{H-H}}$ $= 1.8$ Hz, H_2C-N], 6.08 [t, ${}^4J_{H-H} = 1.8$ Hz, 1H, $=CHC(O)$], 7.34 (t, $J_{\text{H-H}} = 7.4$ Hz, 2H, 6'-H and 7'-H), 7.56 (d, $J_{\text{H-H}} = 7.8$ Hz, 2H, 5'-H and 8'-H), 8.42 ppm (s, 1H, 4'-H); ¹³C-NMR (DMSO⁶): δ 163.2, 162.6 (2C=O), 152.8 (9'-C), 145.6 $(3'-C—C=), 140.1 (4'-C), 132.7, 127.1, 125.8, 118.8, 114.2)$ (7'-C, 8'-C, 5'-C, 6'-C, 10'-C), 113.7 (3'-C), 112.6 $[=CHC(O)],$ 62.6 (OCH₂, 56.5 (H₂C-N₃), 15.6 ppm (OC.CH₃); ms: m/z (EI): 300 (11) [M⁺+1], 299 (22) [M⁺], 257 (100), 242 (18), 225 (35), 198 (25), 171 (16), 142 (10), 114 (8); Anal. Calcd for C₁₅H₁₃N₃O₄ (299.2): C, 60.20; H, 4.38; N, 14.04. Found: C, 60.27; H, 4.43; N, 14.13.

(2Z)-4-Azido-3-(2-oxo-2H-chromen-3-yl)but-2-enenitrile (11c). This was obtained as yellow leaflets, 669 mg $(76%)$; mp 148–150°C (EtOH); IR: v_{max} CN 2216, N₃ 2102, C=O (lactone) 1721, C=C (exocyclic) 1607 cm⁻¹; ¹H-NMR (DMSO⁶): δ 4.44 (d, ⁴J_{H-H} = 2.1 Hz, 2H, *H*₂C-N₃), 6.26 (t, ⁴J_{H-H} = 2.1 Hz, 1H, =CHCN), 7.33 (t, $J_{\text{H-H}} = 7.4$ Hz, 2H, 6'-H and 7'-H), 7.55 (d, $J_{\text{H-H}} = 7.8$ Hz, 2H, 5'-H and 8'-H), 8.45 ppm (s, 1H, 4'-H); ¹³C-NMR (DMSO⁶): δ 161.2 (C=O), 154.7 (9'-C), 148.6 (C=C-CN), 139.7 (4'-C), 132.3, 130.5, 126.4, 118.7, 114.2 (7'-C, 8'-C, 5'-C, 6'-C, 10'-C), 120.4 (3'-C), 115.6 (CN), 98.6 (=CHCN), 52.3 ppm (H₂C-N₃); ms: m/z (EI): 253 (11) $[M^+ + 1]$, 252 (16) $[M^+]$, 224 (77), 210 (48), 197 (100), 169 (85), 140 (42), 114 (13); Anal. Calcd for $C_{13}H_8N_4O_2$ (252.2): C, 61.90; H, 3.20; N, 22.21. Found: C, 61.96; H, 3.24; N, 22.16.

Reaction of 1 with phosphonium salts 13a,b. Preparation of 14a and 15. A dimethylformamide (DMF) solution of 3.5 mmol methyltriphenylphosphonium bromide 13a (or benzyltriphenylphosphonium chloride, 13b) and the azide 1 (0.8 g, 3.5 mmol) was treated with aqueous LiOH. The reaction mixture was stirred at r.t. for 24 h. The product mixture was concentrated to 10 mL, diluted with 30 mL of dist H_2O , acidified with conc. HCl, and then extracted with two portions of ethyl acetate. The AcOEt extracts were combined and washed with 50 mL of dist H_2O and dried, and the solvents were evaporated to dryness. The residue was purified by column chromatography with n -hexane /AcOEt) as the eluents, whereupon compounds $14a$ and 15 , were obtained. Ph₃PO was also isolated.

 $3-I1-(Azidomethyl)vinvl-2H-chromen-2-one$ (14a). This was obtained as yellow leaflets, 635 mg (80%); mp 173-175 $\rm{^{\circ}C}$ (acetone); IR: v_{max} C-N₃ 2106, C=O (lactone) 1720, C=C (exocyclic) 1618 cm⁻¹; ¹H-NMR (DMSO⁶): δ 4.02 (t, $J_{\text{H-H}} = 2.2$ Hz, 2H, H_2CN_3), 5.45, 5.66 (2t(br), 2 \times 1H, $=CH_2$), 7.28 (t, $J_{\text{H-H}} = 7.4 \text{ Hz}, 2\text{H}, 6'$ -H and 7'-H), 7.54 (d, $J_{\text{H-H}} = 7.8 \text{ Hz},$ 2H, 5'-H and 8'-H), 8.43 ppm (s, 1H, 4'-H); ¹³C-NMR $(DMSO⁶)$: δ 156.2 (2'-C=O), 154.2 (9'-C), 139.7 (C=CH₂), 132.3, 129.5, 128.4, 126.4, 118.7, 117.2 (7'-C, 4'-C, 5'-C, 6'- C, 8'-C, 10'-C), 120.4 (3'-C), 114.2 (C=CH₂), 60.2 ppm (H_2C-N_3) ; ms: m/z (EI): 228 (11) $[M^+ + 1]$, 227 (16) $[M^+]$, 185 (44), 167 (48), 149 (100), 99 (65), 77 (28); Anal. Calcd for $C_{12}H_9N_3O_2$ (227.2): C, 63.43; H, 3.99; N, 18.49. Found: C, 63.51; H, 3.96; N, 18.43.

3-[(E,Z)-2-Azido-1-benzylvinyl]-2H-chromen-2-one (15). This was obtained as yellow leaflets, 890 mg (84%); mp 198– 200°C (CHCl₃, mixture of isomers E and Z, 50:50); IR: v_{max} $C-N₃$ 2108, $C=O$ (lactone) 1721, $C=C$ (exocyclic) 1617 cm⁻¹; ¹H-NMR (DMSO⁶): *E* isomer: δ 3.38 (d, 2H, $J_{\text{H-H}}$ = 2.1 Hz, CH₂), 7.18 (t, $J_{H-H} = 2.1$ Hz, 1H, $=$ CHN₃), 7.24–7.38 (m, 3H, H-Ph and H-Ar, E, Z), 7.45–7.67 (m, 6H, H-Ph and H-Ar, E, Z), 8.46, 8.54 ppm (2s, 2 \times 1H, 4'-H, E, Z); Z isomer: δ 3.47 (d, 2H, $J_{\text{H-H}} = 2.2$ Hz, CH₂), 7.23 (t, $J_{\text{H-H}} = 2.2$ Hz, 1H, $=CHN_3$); ms: m/z (EI): 304 (13) [M⁺+1], 303 (21) $[M^+]$, 275 (100), 256 (77), 210 (21), 149 (44), 137 (36), 111 (13), 77 (58); Anal. Calcd for $C_{18}H_{13}N_3O_2$ (303.3): C, 71.28; H, 4.32; N, 13.85. Found: C, 71.33; H, 4.37; N, 13.90.

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REFERENCES AND NOTES

[1] Chimenti, F.; Bolasco, A.; Secci, D.; Bizzarri, B.; Chimenti, P.; Granese, A.; Carradori, S.; Bishay, D. W.; Sayyed, S. M.; Abd El-Hafez, M. A.; Achenbach, H.; Desoky, E. K. J Heterocycl Chem 2010, 47, 729.

[2] Haensch, C.; Hoeppener, S.; Schubert, U. S. Nanotechnology 2008, 19, 35703.

[3] For recent publications in the last two years see ref. 3-12: Abdou, W. M.; Ganoub, N. A.; Geronikaki, A.; Sabry, E. Eur J Med Chem (EJMDCH); 2008; 43: 1015–1024.

[4] Abdou, W. M.; Elnagdy, M. H.; Sadek, K. U.; Abdou, W. A. The Art of Structural Elucidation via Spectroscopy; Dar: Kuwait, 2008; pp 75–108.

[5] Abdou, W. M.; Sediek, A. A.; Khidre, M. D. Monatsh fur Chem 2008, 139, 617.

[6] Abdou, W. M.; Shaddy, A. A. Lett Org Chem (LOC) 2008, 5, 569.

[7] Abdou, W. M.; Khidre, M. D.; Khidre, R. E. J Heterocycl Chem 2008, 45, 1571.

[8] Abdou, W. M.; Shaddy, A. A. Arkivoc 2009, 9, 143.

[9] Abdou, W. M.; Khidre, M. D.; Khidre, R. E. Eur J Med Chem (EJMECH) 2009, 44, 526.

[10] Abdou, W. M.; Shaddy, A. A.; Sediek, A. A. J. Chem Res 2009, 8.

[11] Abdou, W. M.; Ganoub, N. A.; Sabry, E. Z Naturforsch 2009, 46b, 1057.

[12] Abdou, W. M.; Khidre, R. E. Monatsh fur Chem 2010, 141, 219.

[13] Abdou, W. M.; Salem, M. A. I.; Sediek, A. A. Heterocycl Commun 1998, 4, 150.

[14] Abdou, W. M.; Sediek, A. A. Tetrahedron 1999, 55, 14777.

- [15] Kusanur, R. A.; Kulkarni, M. V. Ind J Chem 2005, 44B, 591.
- [16] Nikolova, R.; Bojilova, A.; Rodios, N. A. Tetrahedron 1998, 54, 14407.

[17] Silverstein, R. M.; Webster, F. X. W.; Kiemle, D. J. Spectrometric Identification of Organic Compounds, 7th ed.; John Wiley: New York, 2005.

- [18] Birkofer, L.; Wegner, P. Chem Ber 1967, 100, 3485.
- [19] Cesare, V.; Lyons, T. M.; Lengyel, I. Synthesis 2002, 1716.

[20] Cadogan, J. I.; Stewart, N. J.; Tweddle, N. J. J. Chem Soc Chem Comm 1978, 182.

[21] Lambert, P. H.; Vaultier, M.; Carrié, R. J Org Chem 1985, 50, 5352.

[22] Boulos, L. S.; El-din, N. K. Tetrahedron 1993, 49, 3871.

[23] Ykman, P.; L'Abbe, G.; Smets, G. Tetrahedron 1971, 27, 845.

[24] Kawasaki, T.; Nonaka, Y.; Uemura, M.; Sakamoto, M. Synthesis 1991, 701.

[25] Kawasaki, T.; Nonaka, Y.; Ohtsuka, H.; Sato, H.; Sakamoto, M. J Chem Soc Perkin Trans 1990, 1, 1101.

[26] Kawasaki, T.; Nonaka, Y.; Sakamoto, M. Heterocycles 2000, 53, 1681.

[27] Mérour, J. Y.; Gadonneix, P.; Andrieu, B. M.; Desarbre, E. Tetrahedron 2001, 57, 1995.